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2-Methylbenzamidoxime

Hai-Bo Wang,* Wei-Lin Ding, Zhi-Tao Xing and Pin-Liang Wang

Department of Applied Chemistry, College of Science, Nanjing University of Technolgy, Xinmofan Road No.5 Nanjing, Nanjing 210009, People's Republic of China

Correspondence e-mail: wanghaibo@njut.edu.cn

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(\text{C-C}) = 0.003 \text{ Å}$ R factor = 0.060 wR factor = 0.189Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

In the crystal structure of the title compound, $C_8H_{10}N_2O$, which is a benzonitrile derivative, an intramolecular $N-H\cdots O$ hydrogen bond and intermolecular $N-H\cdots O$ and $O-H\cdots N$ hydrogen bonds are present.

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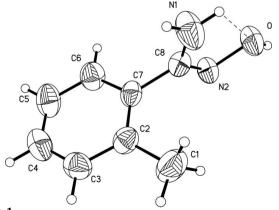
Comment

As part of our ongoing studies of benzonitrile derivatives, we report here the crystal structure of the title compound, (I).

The geometrical parameters for (I) are normal; its molecular structure is shown in Fig. 1. The dihedral angle between the mean planes of the C2–C7 benzene ring and the C7/C8/N1/N2/O group is 81.68 (11)°. A combination of intramolecular and intermolecular hydrogen bonds (Table 2) helps to establish the crystal packing.

Experimental

Three solutions were prepared: 2-methylbenzonitrile (20 mmol) in ethanol (8 ml), hydroxylamine hydrochloride (20 mmol) in ethanol (6 ml) and potassium carbonate (10 mmol) in water (10 ml). The three solutions were mixed and the resulting mixture was refluxed for



The molecular structure of (I), showing displacement ellipsoids drawn at the 30% probability level (arbitrary spheres for the H atoms). The dashed line indicates the intramolecular $N-H\cdots O$ hydrogen bond.

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24 h. After cooling and filtration, crude compound (I) was obtained. Pure compound (I) was obtained by crystallization from a mixture of ethanol (6 ml) and water (2 ml). Yellow crystals of (I) suitable for X-ray diffraction were obstained by slow evaporation of an ethanol solution.

Crystal data

$C_8H_{10}N_2O$	Z = 4		
$M_r = 150.18$	$D_x = 1.218 \text{ Mg m}^{-3}$		
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation		
a = 9.751 (2) Å	$\mu = 0.08 \text{ mm}^{-1}$		
b = 7.0880 (14) Å	T = 293 (2) K		
c = 11.982 (2) Å	Block, yellow		
$\beta = 98.60 \ (3)^{\circ}$	$0.40 \times 0.20 \times 0.20 \text{ mm}$		
$V = 818.8 (3) \text{ Å}^3$			

Data collection

Enraf-Nonius CAD-4	1600 independent reflections		
diffractometer	1124 reflections with $I > 2\sigma(I)$		
$\omega/2\theta$ scans	$R_{\rm int} = 0.017$		
Absorption correction: ψ scan	$\theta_{\rm max} = 26.0^{\circ}$		
(North et al., 1968)	3 standard reflections		
$T_{\min} = 0.968, T_{\max} = 0.984$	every 200 reflections		
1694 measured reflections	intensity decay: none		

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.1125P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.060$	+ 0.0887 <i>P</i>]
$wR(F^2) = 0.189$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} < 0.001$
1600 reflections	$\Delta \rho_{\text{max}} = 0.20 \text{ e Å}^{-3}$
100 parameters	$\Delta \rho_{\min} = -0.22 \text{ e Å}^{-3}$
H-atom parameters constrained	

Table 1 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-H\cdots A$
$ \begin{array}{c} N1 - H1B \cdots O^{i} \\ N1 - H1A \cdots O \\ O - H1w \cdots N2^{ii} \end{array} $	0.86	2.44	3.267 (3)	163
	0.86	2.22	2.542 (3)	102
	0.82	2.08	2.787 (3)	145

Symmetry codes: (i) $-x + \frac{3}{2}$, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) -x + 1, -y, -z.

All H atoms were positioned gemoetrically (C-H = 0.93-0.97, O-H = 0.82, N-H = 0.86 Å) and refined as riding with $U_{\rm iso}$ (H) = 1.2 or 1.5 $U_{\rm co}$ of the carrier atom.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

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